

# On the SCF theory of continuum solvent effects representation: Introduction of local dielectric effects

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The introduction of local dielectric effects within the SCF theory of continuum solvent effects representation is examined at a semiempirical level. The formalism is developed in the frame of the reaction field theory within the effective charge approximation. The solvation free energies of  $\text{Li}^+$ ,  $\text{Na}^+$ ,  $\text{F}^-$ , and  $\text{Cl}^-$  ions in water were calculated in order to illustrate the reliability of the proposed model. The extension to molecules and molecular ions was performed including a desolvation corrective term related to the specific neighborhood of each atomic center. The results show a qualitative agreement with experimental data. A comment on the solvationlike models for incorporating the solvent effect into the Hamiltonian is also given. Copyright © 1985 John Wiley & Sons, Inc.