

# Benchmark values of chemical potential and chemical hardness for atoms and atomic ions (including unstable anions) from the energies of isoelectronic series

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## Resumen

We present benchmark values for the electronic chemical potential and chemical hardness from reference data for ionization potentials and electron affinities. In cases where the energies needed to compute these quantities are not available, we estimate the ionization potential of the metastable (di) anions by extrapolation along the isoelectronic series, taking care to ensure that the extrapolated data satisfy reasonable intuitive rules to the maximum possible extent. We also propose suitable values for the chemical potential and chemical hardness of zero-electron species. Because the values we report are faithful to the trends in accurate data on atomic energies, we believe that our proposed values for the chemical potential and chemical hardness are ideally suited to conceptual studies of chemical trends across the periodic table. The critical nuclear charge ( $Z$  critical) of the isoelectronic series with  $2 < N < 96$  has also been reported for the first time.

## Palabras clave

**KeyWords Plus:** DENSITY-FUNCTIONAL THEORY; HARD/SOFT-ACID/BASE PRINCIPLE; FRONTIER-ELECTRON THEORY; MAXIMUM HARDNESS; SOFT ACIDS; ELECTRONEGATIVITY EQUALIZATION; HSAB PRINCIPLE; GROUND-STATE; VARIATIONAL-PRINCIPLES; PERTURBATION-THEORY

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