

Theoretical study on interactions of fluorinated organomercurials with arene and gold fragments

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This journal is © the Owner Societies. The electronic structure and spectroscopic properties of

$[\text{Hg}(\text{C}_6\text{F}_5)_2]_2\text{-}\{\text{L}\}$,

$[\text{Hg}_3(\text{o-C}_6\text{F}_4)_3]_2\text{-}\{\text{L}\}$ (L = naphthalene,

biphenyl, fluorene) and

$[\text{Hg}_3(\text{o-C}_6\text{F}_4)_3]\{\text{Au}_3(?-\text{C}(\text{OEt})=\text{NC}_6\text{H}_4\text{CH}_3)_n$ (n = 1, 2) adducts were studied at the HF,

MP2, SCS-MP2, DFT and DFT-D3 levels. The intermolecular interactions among the fragments

were analyzed using the levels of calculations proposed. The energy decomposition analysis at the

TPSS-D3 level was used to define the dominant components of the interaction. The van der Waals

interactions between mercury and arene (Hg-arene) were found to be the main short-range stability contribution in the $[\text{Hg}(\text{C}_6\text{F}_5)_2]_2\text{-}\{\text{L}\}$ and

$[\text{Hg}_3(\text{o-C}_6\text{F}_4)_3]_2$ and

$[\text{Hg}_3(\text{o-C}_6\text{F}_4)_3]_2$