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# REVISED STRUCTURES OF THE AZAFLUORENONE ALKALOIDS FROM GUATTERIA DIELSIANA\*

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Key Word Index—Guatteria dielsiana; Annonaceae; structural revision; synthesis; onychine; 6-methoxyonychine; dielsine; dielsinol.

Abstract—The structures of three alkaloids isolated from *Guatteria dielsiana*, 6-methoxyonychine, dielsine and dielsinol, originally described as 1-azafluoren-9-ones, are revised to 4-azafluoren-9-ones on the basis of previous synthetic work and current studies on alkaloids of the Annonaceae. Synthetic 6-methoxyonychine (6-methoxy-1-methyl-4-azafluoren-9-one) is shown to be identical with the natural product.

### INTRODUCTION

The highly conjugated, yellow alkaloid onychine (1) was first isolated in 1976 from Onychopetalum amazonicum, when its structure was misassigned as 4-methyl-1azafluoren-9-one [1]. 1-Methyl-4-azafluoren-9-one had been synthesized previously [2], was prepared again by another route a year later [3], and in 1979 was synthesized once more and shown to agree with onychine in its spectral properties and those of its borohydride reduction product, while the 4-methyl-1-aza isomer was proved to be different [4]. In spite of these results, the erroneous formula was retained several years later when this substance was reisolated from Cleistopholis patens [5] and Guatteria dielsiana [6], and was invoked as the framework upon which the structures of several novel, more oxygenated azafluorenones from the latter plant were based [6]. More recently, four other related alkaloids have been isolated from Meiogyne virgata [7] and Oxandra cf. major [8], and thorough NMR studies of these compounds leave no doubt that they are 1-methyl-4-azafluoren-9-one derivatives. A biogenetic scheme has also been suggested which, among other things, points to a possible route for the formation of 4-azafluoren-9-ones, but not the 1-aza isomers [7].

#### **RESULTS AND DISCUSSION**

In this context it becomes necessary to revise the structures of the Guatteria dielsiana azafluorenone derivatives: 6-methoxyonychine (2), dielsine (3) and dielsinol (4) [6]. All three are now depicted as sharing the 1-methyl-4-azafluorene skeleton, in line with the correct structures of onychine and of its more recently discovered analogues. In support of this revision, the O-crotyloximes (mixture of geometric isomers) of 6-methoxyindanone were heated under air at 180° to give a complex mixture of products from which 6-methoxy-1-methyl-4-azafluoren-9-one was isolated chromatographically. This approach is unambiguous as regards the position of the substituent on the benzene ring [4]. Comparison of synthetic 6-methoxy-1methyl-4-azafluoren-9-one (2) with 6-methoxyonychine from Guatteria dielsiana showed them to be identical in all respects.

#### EXPERIMENTAL

Synthesis of 6-methoxy-1-methyl-4-azafluoren-9-one (2). A mixture of 6-methoxyindanone [9] (1.8 g), O-crotylhydroxylamine [10] (1.6 g), NaOAc (0.60 g), Na<sub>2</sub>CO<sub>3</sub> (1.0 g) and EtOH (15 ml) was refluxed for 2 hr. The solvent was removed and the residue was extracted with CH<sub>2</sub>Cl<sub>2</sub> to afford a mixture of (*E*)- and (*Z*)oximes (2.8 g). The oximes (1.0 g) were heated under air at 170–180° for 20 hr, the reaction products were taken up in CH<sub>2</sub>Cl<sub>2</sub>, the basic constituents extracted with 4% HCl, the aq. soln made alkaline with NH<sub>3</sub> and extracted with CH<sub>2</sub>Cl<sub>2</sub>, giving a mixture of 1- and 3-methyl-4-azafluorenes and traces of 4azafluoren-9-ones (0.12 g). This was fractionated by 'flash' chromatography and TLC on silica gel, eluting with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (99:1), to yield 6-methoxy-1-methyl-4-azafluoren-9-one as a yellow gum (5 mg): EIMS m/z (rel. int.) 225.0795 [M<sup>+</sup>] (100),

<sup>\*</sup>Part 79 in the series "Alcaloïdes des Annonacées". For part 78 see S. Rasamizafy, R. Hocquemiller, A. Cavé and A. Fournet, J. Nat. Prod. (submitted).

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C<sub>14</sub>H<sub>11</sub>NO<sub>2</sub> calc. 225.0790, 224 [M<sup>+</sup> -H] (10.8), 210 [M<sup>+</sup> -CH<sub>3</sub>] (9.6), 197 [M<sup>+</sup> -CO] (1.4), 196 [M<sup>+</sup> -H -CO] (11.4), 195 [M<sup>+</sup> -CH<sub>2</sub>O] (13.6), 182 (7.6), 167 (6.2), 154 (7.3), 127 (8.8), 86 (12.5), 84 (22.5), 49 (32.9); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  2.63 (s, 3H, C-1-CH<sub>3</sub>), 3.94 (s, 3H, OCH<sub>3</sub>), 6.87 (dd, J<sub>o</sub> = 8.2 Hz, J<sub>m</sub> = 2.3 Hz, 1H, H-7), 6.96 (d, J = 5.3 Hz, 1H, H-2), 7.35 (d, J<sub>m</sub> = 2.3 Hz, 1H, H-5), 7.64 (d, J<sub>o</sub> = 8.2 Hz, 1H, H-8), 8.39 (d, J = 5.3 Hz, 1H, H-3); IR  $\nu$  film cm<sup>-1</sup>: 1700, 1610, 1595 sh, 1570, 1555 sh, 1468, 1433, 1425 sh, 1386, 1357, 1272 sh, 1260, 1245 sh, 1213, 1180, 1097, 1085 sh, 1048, 1010, 928, 892, 872, 830, 792, 767; UV  $\lambda$  EiOH + m (log  $\varepsilon$ ): 214 (4.09), 225 (4.36), 235 (4.36), 245 (4.36), 276 sh (4.36), 280 (4.21), 292 (4.36), 326 (3.44), 340 (3.44);  $\lambda$  EiOH + HCl nm (log  $\varepsilon$ ) 216 (4.05), 230 sh (4.09), 240 sh (4.21), 248 (4.26), 276 sh (4.09), 283 (4.11), 294 (4.09), 308 sh (3.74), 328 (3.36), 343 (3.36).

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