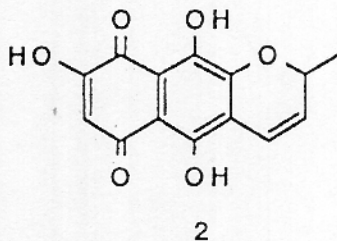
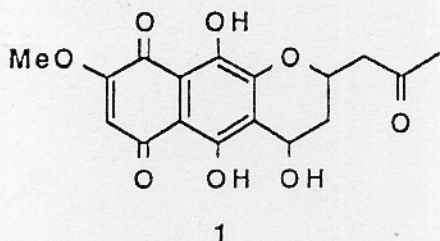


A FACILE SYNTHESIS OF THE PYRANONAPHTHAZARIN SYSTEM

Claudio Saitz B., Jaime A. Valderrama, and Ricardo Tapia *
Facultad de Química. Pontificia Universidad Católica de Chile
Casilla 306, Santiago 22, Chile

ABSTRACT :The synthesis of pyranoquinone **9** starting from hydroxyquinone **4** through Michael adduct **5** is described.

The synthesis of pyranonaphthoquinones have received increasing attention because of the interesting biological properties of these heterocyclic quinones either from natural or synthetic origin¹. The research in this area has been focused mainly on the synthesis of pyranoquinones derived from 1,4-naphthoquinone or 5-hydroxy-1,4-naphthoquinone², but those derived from 5,8-dihydroxy-1,4-naphthoquinone (naphthazarin) such as erythrostominone **13** and the pyranonaphthazarin **24** have been unexplored.



In our continued effort to find a synthetic route to obtain highly oxygenated pyranoquinones, we developed an efficient method to synthesize pyranonaphthoquinones using Michael adducts of hydroxyquinones⁵. In this study we want to extend this approach to the synthesis of pyranonaphthazarin **9**.

*To whom correspondence should be addressed

The Michael adduct approach to pyranoquinones requires an hydroxyquinone, and in order to explore this route to obtain pyranonaphthazarins we chose the known hydroxyquinone **4**⁶ as starting material.

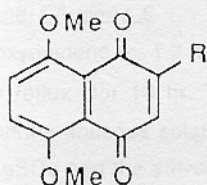
Acid-catalyzed addition of water to 5,8-dimethoxy-1,4-naphthoquinone **3**⁷ gave hydroxyquinone **4** in 90% yield. Reaction of **4** with methyl vinyl ketone (MVK) in pyridine-*t*-butanol mixture afforded Michael adduct **5** in 70% yield. The IR spectrum of **5** showed two carbonyl absorptions at 1660 and 1640 cm^{-1} and a singlet at δ 2.17 ppm was observed by ¹Hnmr confirming its structure.

Attempts to convert adduct **5** into the corresponding alcohol **6**, by reduction with sodium borohydride, indicated that **6** underwent a ready cyclization to *o*-pyranoquinone **7** during the isolation procedure. Since compound **7** could provide a pyranonaphthazarin by acid-catalyzed rearrangement⁸, we decided to induce the cyclization of alcohol **6** leaving it overnight in the chloroform extract, under aerobic condition. This procedure afforded quinone **7** in 92% isolated yield.

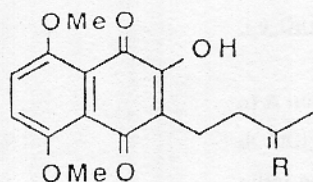
The conversion of *o*-pyranonaphthoquinones to *p*-pyranonaphthoquinones was studied in detail by Ettlinger⁸, and in our case the isomerization of the heterocyclic quinone **7** was achieved by heating in 20% sulfuric acid for 30 minutes to afford *p*-pyranoquinone **8** in 95% yield.

Demethylation of **8** by treatment with aluminum chloride at room temperature yielded pyranonaphthazarin **9** or **10** in good yield (71%). The product displayed the signals of two coupled protons ($J = 9.5$ Hz) at δ 7.16 and 7.26 ppm for the naphthazarin system. Considering this fact and, the well known study reported by Moore and Scheuer⁹ on the influence of a variety of substituents on the tautomerism of naphthazarins, it is highly probable that the naphthazarin released from **8** has the pyran ring fused to the quinonic nucleus, as in structure **9**. Interestingly, quinone **9** can be obtained in one step from **7** by heating the latter in 20% sulfuric acid for 2 hr in 77% yield.

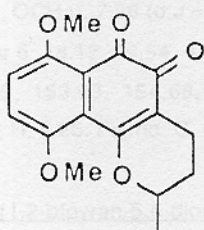
In conclusion, we have shown that our method is useful to obtain highly oxygenated pyranoquinones. We are currently studying the synthesis of erythrostominone derivatives using this approach.



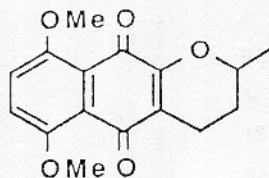
3 R = H
4 R = OH



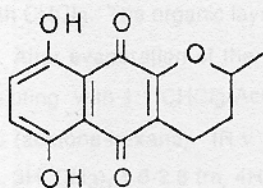
5 R = O
6 R = H, OH



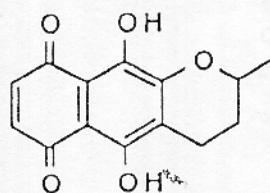
7



8



9



10

EXPERIMENTAL

Melting points were determined with a Koller modified apparatus and are not corrected. IR spectra were recorded on a Perkin-Elmer Model 1310 Spectrometer. ^1H and ^{13}C nmr spectra were recorded on Varian XL-100 and Varian XL-300 spectrometers, respectively, using tetramethylsilane as internal reference. Column chromatography was performed on Silica-gel 60 (Merck). Elemental analyses were performed at the Instituto de Química Orgánica General, Madrid, Spain.

5,8-dimethoxy-2-hydroxynaphthalene-1,4-dione 4.

Obtained in 90% yield from 5,8-dimethoxynaphthalene-1,4-dione **3**⁷ using Fariña et al.⁶ procedure.

5,8-Dimethoxy-2-hydroxy-3-(3-oxobutyl)naphthalene-1,4-dione 5.

A mixture of 500 mg (2.14 mmol) of hydroxyquinone 4, 1.2 mL of MVK and 50 mL of pyridine-*t*-butanol (1:5) was heated under reflux for 14 hr. The residue after removal of solvent was acidified with 5% HCl, and the aqueous solution was extracted with CH₂Cl₂. The organic layer was dried over MgSO₄ and the solvent removed under vacuum. The crude product was chromatographed on silica gel using 1:1 chloroform-ethyl acetate as eluent to give 460 mg (71%) of quinone 5, mp 141-143 °C (acetone-hexane). IR ν max : 3240 (OH), 1660 and 1640 (C=O) cm⁻¹. ¹H-nmr δ : 2.17 (s, 3H, CH₃), 2.6-2.9 (m, 4H, CH₂CH₂), 3.97 (s, 3H, OCH₃), 4.00 (s, 3H, OCH₃), 7.26 (d, J = 9.5 Hz, 1H), 7.41 (d, J = 9.5 Hz, 1H), 7.60 (s, 1H, OH) ppm. ¹³C-nmr δ : 18.12, 29.54, 41.73, 56.74, 57.12, 118.92, 121.83, 122.98, 123.02, 123.04, 152.05, 153.93, 154.68, 179.99, 184.38, 207.89 ppm. Anal. Calcd. for C₁₆H₁₆O₆: C, 63.16; H, 5.26. Found: C, 63.03; H, 5.24.

3,4-Dihydro-7,10-dimethoxy-2-methyl-2H-naphtho[1,2-b]pyran-5,6-dione 7.

To an ice-cold solution containing 320 mg (1.05 mmol) of quinone 5 in 35 mL of 10:1 THF/MeOH was added in portions 111 mg (3.15 mmol) of NaBH₄ with stirring. The mixture was stirred for 50 min in an ice bath, then quenched with 50 mL of water. The solution was acidified with 5% HCl and extracted with CHCl₃. The organic layer was dried over MgSO₄ and then allowed to stand overnight. After evaporation of the solvent, the residue was purified by column chromatography eluting with 1:1 CHCl₃/AcOEt to yield 280 mg (97%) of pyranoquinone 7, mp 174-176 °C (acetone-hexane). IR ν max : 1670 and 1620 (C=O) cm⁻¹. ¹H-nmr δ : 1.50 (d, J = 6 Hz, 3H, CH₃), 1.6-2.8 (m, 4H, CH₂CH₂), 3.82 (s, 3H, OCH₃), 3.90 (s, 3H, OCH₃), 4.2-4.4 (m, 1H, CH), 7.08 (d, J = 9.5 Hz, 1H), 7.28 (d, J = 9.5 Hz, 1H) ppm. ¹³C-nmr δ : 18.49, 20.56, 27.44, 56.66, 58.62, 75.09, 112.90, 116.91, 119.11, 121.99, 125.45, 151.69, 156.75, 165.53, 179.45, 180.60 ppm. Anal. Calcd. for C₁₆H₁₆O₅: C, 66.67; H, 5.56. Found: C, 66.38; H, 5.40.

3,4-Dihydro-6,9-dimethoxy-2-methyl-2H-naphtho[2,3-b]pyran-5,10-dione 8.

A solution of 40 mg (0.14 mmol) of quinone 7 in 50 mL of 20% H₂SO₄ was heated at 50 °C for 30 min with stirring. The solution was extracted with ether and the extract was dried over MgSO₄, and evaporated in vacuo. The residue was chromatographed on silica gel using 1:1 CHCl₃/AcOEt as eluent to give 38 mg (95%) of pyranoquinone 8, mp 160-162 °C (acetone-hexane). IR ν max : 1650 and 1620 (C=O) cm⁻¹. ¹H-nmr δ : 1.46 (d, J = 6 Hz, 3H, CH₃), 1.6-2.7 (m, 4H, CH₂CH₂), 3.94 (s, 3H, OCH₃), 3.95 (s, 3H, OCH₃), 4.1-4.4 (m, 1H, CH), 7.20 (d, J = 9.5 Hz, 1H), 7.30 (d, J = 9.5 Hz, 1H) ppm. Anal. Calcd. for C₁₆H₁₆O₅: C, 66.67; H, 5.56. Found: C, 66.40; H, 5.81.

3,4-Dihydro-6,9-dihydroxy-2-methyl-2H-naphtho[2,3-b]pyran-5,10-dione 9.

a) A mixture of 28 mg (0.1 mmol) of quinone 8, 5 mL of CH_2Cl_2 and 100 mg (0.76 mmol) of AlCl_3 was stirred at room temperature for 6 hr. After cooling in an ice bath, water and ether were added and the layers separated. The aqueous layer was extracted with ether and the combined organic layers were dried over MgSO_4 . The solvents were evaporated in vacuo and the residue was chromatographed on silica gel using CHCl_3 as eluent to give 18 mg (71%) of pyranoquinone 9, mp 128-130 °C. IR ν max : 1600 (C=O) cm^{-1} . $^1\text{H-nmr}$ δ : 1.52 (d, $J = 6$ Hz, 3H, CH_3), 1.2-4.0 (m, 4H, CH_2CH_2), 4.2-4.6 (m, 1H, CH), 7.16 (d, $J = 9.5$ Hz, 1H), 7.26 (d, $J = 9.5$ Hz, 1H), 12.20 (s, 1H, OH), 12.70 (s, 1H, OH) ppm. $^{13}\text{C-nmr}$ δ : 17.95, 20.36, 27.04, 74.75, 110.56, 111.25, 121.86, 127.96, 130.14, 155.86, 157.89, 182.39, 187.64 ppm. Anal. Calcd. for $\text{C}_{14}\text{H}_{12}\text{O}_5$: C, 64.61; H, 4.61. Found: C, 64.50; H, 4.27.

b) A mixture of 100 mg (0.35 mmol) of quinone 7 in 25 mL of 20% H_2SO_4 was heated to reflux for 2 hr. The aqueous mixture was extracted with ether and the extract was washed with brine, dried, and evaporated in vacuo. The residue was chromatographed on silica gel using CHCl_3 as eluent to give 54 mg of pyranoquinone 9. Further elution with the same solvent afforded 22 mg of the starting quinone 7. Based on recovered 7 the yield of 9 is 77%.

ACKNOWLEDGEMENTS

We thank "Fondo Nacional de Desarrollo Científico y Tecnológico" (Research Grant 376-89) for financial support, and "Fundación Andes" for a fellowship to C. Saitz.

REFERENCES AND NOTES

1. a) Wu, T. S., Tien, H. J., Yeh, M. Y. and Lee, K.H. *Phytochemistry*, 1988, 27, 3787. b) Hayashi, T., Smith, F. T., and Lee, K.H., *J. Med. Chem.*, 1987, 30, 2005. c) Berdy, J., "Handbook of Antibiotic Compounds", Vol III, CRC Press, Florida, 1980, p 221.
2. a) Gupta, R. B., and Franck, R. W., *J. Am. Chem. Soc.*, 1989, 111, 7668. b) Brade, W., and Vasella, A., *Helv. Chim. Acta*, 1989, 72, 1649. c) Matsumoto, T., Ichihara, A.,

- Yanagiya, M., Yuzawa, T., Sannai, A., Oikawa, H., Sakamura, S., and Eugster, C. H., *Helv. Chim. Acta*, 1985, 68, 2324. d) *Houben-Weyl Methoden der Organischen Chemie, Chinon-I*; George Thieme Verlag, Berlin, 1977; Vol VII-3a, p. 373.
3. Cross, B. E., and Zammitt, L. J., *J. Chem. Soc. Perkin Trans. 1*, 1972, 380.
 4. Moore, R. E., Singh, H., and Scheuer, P. J., *Tetrahedron Lett.*, 1968, 4581.
 5. a) Saitz, C., Valderrama, J. A., and Tapia, R., *Synth. Commun.*, 1990, 20, 3103. b) Tapia, R. and Valderrama, J. A., *Bol. Soc. Chil. Quím.* 1990, 35, 271.
 6. Fariña, F., Martínez-Utrilla, R. and Paredes, M. C., *Synthesis*, 1981, 300.
 7. Terashima, S., Kawasaki, M. and Matsuda, F., *Tetrahedron*, 1988, 44, 5713.
 8. Ettliger, M. G., *J. Am. Chem. Soc.*, 1950, 72, 3090.
 9. Moore, R. E., and Scheuer, P. J., *J. Org. Chem.*, 1966, 31, 3272.