

# Neuroprotective Effect of a New 7,8-Dihydroxycoumarin-Based Fe<sup>2+</sup>/Cu<sup>2+</sup> Chelator in Cell and Animal Models of Parkinson's Disease

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© 2016 American Chemical Society. Disturbed iron homeostasis, often coupled to mitochondrial dysfunction, plays an important role in the progression of common neurodegenerative diseases such as Parkinson's disease (PD). Recent studies have underlined the relevance of iron chelation therapy for the treatment of these diseases. Here we describe the synthesis, chemical, and biological characterization of the multifunctional chelator 7,8-dihydroxy-4-((methylamino)methyl)-2H-chromen-2-one (DHC12). Metal selectivity of DHC12 was Cu<sup>2+</sup> > Fe<sup>2+</sup> > Zn<sup>2+</sup> > Fe<sup>3+</sup>. No binding capacity was detected for Hg<sup>2+</sup>, Co<sup>2+</sup>, Ca<sup>2+</sup>, Mn<sup>2+</sup>, Mg<sup>2+</sup>, Ni<sup>2+</sup>, Pb<sup>2+</sup>, or Cd<sup>2+</sup>. DHC12 accessed cells colocalizing with Mitotracker Orange, an indication of mitochondrial targeting. In addition, DHC12 chelated mitochondrial and cytoplasmic labile iron. Upon mitochondrial complex I inhibition, DHC12 protected plasma membrane and mitochondria against lipid peroxidation, as detected by the reduced formation of 4-hydroxynonenal adducts an