

Ab initio crystal structure determination of Na₂Si₃O₇ from conventional powder diffraction data

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The crystal structure of Na₂Si₃O₇ has been determined by direct methods using integrated intensities of conventional X-ray powder diffraction data and subsequently refined with the Rietveld technique. The title compound was prepared from Na₂Si₃O₇ × H₂O by careful thermal decomposition at 440°C. Sodium trisilicate adopts monoclinic symmetry, space group P2₁/c with unit cell parameters $a = 7.1924(5) \text{ \AA}$, $b = 10.6039(8) \text{ \AA}$, $c = 9.8049(7) \text{ \AA}$, $\beta = 120.2478(4)^\circ$, $V = 646.0(9) \text{ \AA}^3$ and $Z = 4$. It belongs to the group of interrupted framework silicates of four- and three-connected [SiO₄]-tetrahedra with a ratio of Q₃:Q₄ = 2:1. Within the framework the sodium atoms are coordinated by 4 to 6 oxygen ligands. The porous character of the new phase is reflected in a framework density $FD = 18.6 \text{ T-atoms}/1000 \text{ \AA}^3$, a value which is comparable to those observed in zeolitic materials. The topology of the tetrahedral network is identical to the one observed in the hydrous sodium silicate Na₂Si₃O₇ × H₂O. Differences