

Analogues of the Cyclic Hydroxamic Acid

2,4-Dihydroxy-7-methoxy-2H-1,4-benzoxazin-3-one: Decomposition to Benzoxazolinones and Reaction with β -Mercaptoethanol

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Analogues of the aglucones of naturally occurring cyclic hydroxamic acids

(2,4-dihydroxy-1,4-benzoxazin-3-ones) from Gramineae (Poaceae) have been synthesized by the reductive cyclization of the ring-substituted methyl β -(*o*-nitrophenoxy)- β -methoxyacetates, followed by demethylation of the C-2 methoxy group with BBr₃ or BCl₃ to reveal the 2-hydroxy group. A structure-activity series was produced by varying the substituent at C-7 on the aromatic ring [R = MeO (1), *t*-Bu (6), Me (7), H (8), Cl (9), F (10), CO₂Me (11a)]. The pK_a values for the hydroxamic acid and the phenol moieties were determined for each member of the C-7 series. They correlated well with ρ in a linear free energy relationship (LFER) yielding values of $\rho = 0.71$ (with ρ_p) for pK_{a1} (the hydroxamic acid) and $\rho = 1.6$ (with ρ_m) for pK_{a2} (the phenol). A LFER also existed between the rate constants for the unimolecular decomposition of these hydroxamic acids to benzoxazolinones and ρ^+ ($\rho^+ = -1.1$). The rates of hydroxamic acid red