

The preparation and x-ray structure of  
chlorobis(N-methyl-N-phenylhydrazido)bis(triphenylphosphine)molybdenum  
trifluoromethanesulfonate hemihydrate:  $[\text{Mo}(\text{NNMePh})_2\text{Cl}(\text{PPh}_3)_2]^+\text{CF}_3\text{SO}_3^- \cdot 1/2\text{H}_2\text{O}$

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$[\text{Mo}(\text{NNMePh})_2\text{Cl}(\text{PPh}_3)_2]^+\text{CF}_3\text{SO}_3^- \cdot 1/2\text{H}_2\text{O}$ ,  $[\text{1}]^+$ , has been synthesized by reaction of silver triflate,  $\text{AgCF}_3\text{SO}_3$ , with  $[\text{Mo}(\text{NNMePh})_2\text{Cl}_2(\text{PPh}_3)_2]$  in  $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{CN}$  and subsequent isolation and recrystallization from a  $\text{CH}_2\text{Cl}_2$  solution layered with n-hexane. This complex has been characterized by  $^1\text{H}$  and  $^{31}\text{P}$  NMR, IR spectroscopy and by single crystal X-ray diffraction analysis. The geometry of the cationic complex,  $[\text{1}]^+$ , exhibits a distorted trigonal bipyramidal with the two triphenylphosphine ligands occupying the apical positions and the equatorial plane defined by a chloro and the  $\alpha$ -nitrogen atoms of the two nearly linear N-methyl-N-phenylhydrazido ligands, NNMePh.