

Three new pseudohalide bridged dinuclear Zn(II) Schiff base complexes: Synthesis, crystal structures and fluorescence studies

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Three new dinuclear Zn(II) complexes $[Zn(L)(\eta^1,1-N_3)Zn(L)(N_3)] \cdot 1.5H_2O$ (1), $[Zn(L)(\eta^1,1-NCO)Zn(L)(NCO)] \cdot 1.5H_2O$ (2) and $[Zn(L)(\eta^1,1-NCS)Zn(L)(NCS)(OH_2)]$ (3) have been synthesized from a potentially tetradentate N_2O_2 -donor Schiff base ligand LH, $[LH = (OCH_3)(OH)C_6H_3CHN(CH_2)_2N(CH_3)_2]$, which is the condensation product of o-vanillin and 2-dimethylaminoethylamine. All the three complexes 1, 2 and 3 have been characterized by elemental analysis, IR and 1H NMR spectroscopy, TGA and fluorescence studies. Finally, their structures have been established by the single crystal X-ray diffraction method. Structural studies reveal that in complexes 1, 2 and 3 the two Zn(II) centers are held together by a η^2 -phenolato oxygen atom and also by an end-on pseudohalide nitrogen (azide for 1; cyanate for 2; thiocyanate for 3) atom. Among the two deprotonated Schiff base ligands present in each complex, one acts as a tetradentate ligand (N_2O_2 donor set) while the other acts as a tridentate ligand (N_2O don