

Reactivity of $[\text{Mo}(\text{NHNPh}_2)(\text{NNPh}_2)\text{Br}_2(\text{acac})]$ toward di-imines. X-ray crystal structure of $[\text{Mo}(\text{NNPh}_2)_2\text{Br}_2(\text{o-phen})]$

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Complex $[\text{Mo}(\text{NHNPh}_2)(\text{NNPh}_2)\text{Br}_2(\text{acac})]$, acac=acetylacetonate, reacts in acetonitrile with 1,10-phenanthroline, o-phen, and 2,2'-bipyridine, bpy, to afford complexes $[\text{Mo}(\text{NNPh}_2)_2\text{Br}_2(\text{o-phen})]$, I, and $[\text{Mo}(\text{NNPh}_2)_2\text{Br}_2(\text{bpy})]$, II. These complexes have been characterized by IR, UV-vis and ^1H NMR spectroscopies, and elemental analysis. The structure of complex I was solved by single crystal X-ray diffraction. The metal center has a distorted octahedral environment in which the diphenylhydrazido(2-) ligands occupy mutually cis positions but trans to the nitrogen atoms of the o-phen ligand. The bromo ligands occupy mutually trans positions.